

Stereoselective synthesis of chiral tertiary alcohol building blocks via neighbouring group participation from tri-substituted olefins

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Abstract

A regio- and stereoselective preparation of chiral quaternary 1,2/1,3-diols from acyclic tri-substituted alkenes is disclosed. Optically active tertiary alcohols are the constituents of several bioactive natural products, pharmaceuticals and a general method for their preparation is desirable. A sulfinyl moiety has been utilized as an intramolecular nucleophile.

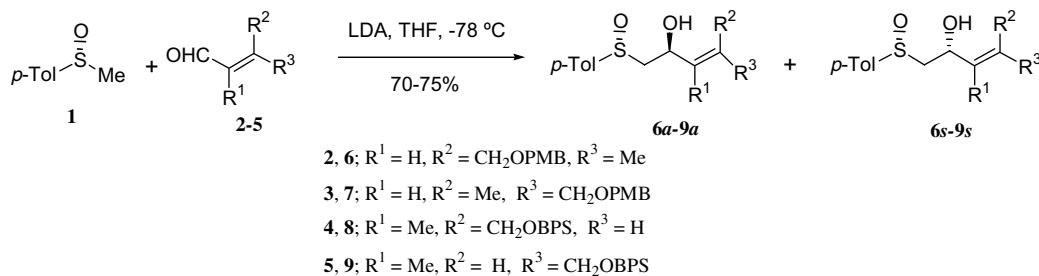
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Keywords: Chiral quaternary 1,2/1,3-diols; Tri-substituted alkenes; Sulfoxide

The introduction of a chiral quaternary carbon centre poses a considerable challenge to synthetic organic chemists.¹ Optically active tertiary alcohols are the constituents of several bioactive natural products,² pharmaceuticals and employed as versatile building blocks. Much effort has therefore been devoted to the synthesis of chiral tertiary alcohols, these include (a) asymmetric oxidation³ (b) desymmetrization of prochiral tertiary alcohols⁴ (c) enantio-/diastereoselective nucleophilic addition to ketones⁵ and (d) stereospecific insertion of carbene into secondary alcohols.⁶ Since most of the reported methods have their limitation in

terms of the substrate diversity, new methods applicable to a variety of substrates are needed. In this context, we report herein, a highly regio- and stereoselective preparation of chiral 1,2/1,3 diols possessing a quaternary stereogenic centre from acyclic tri-substituted allyl alcohols.⁷ We also present the utility of one of the building blocks for the synthesis of C7–C11 fragment of fostriecin.

The olefinic substrates **6–9** were prepared as an equimolar mixture of epimers by condensing the lithium anion of methyl *p*-tolyl sulfoxide⁸ **1** and unsaturated aldehydes⁹ **2–5**, Scheme 1. The *anti*-(**6a–9a**) and *syn*-isomers¹⁰ (**6s–9s**) were



Scheme 1.

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Table 1
 Regio- and stereoselective chiral tertiary alcohol preparation^a

S. No	Substrate	Product		Yield % (<i>Syn:Anti</i>)	Derivative
		<i>Syn</i> C2,C3	<i>Anti</i> C2,C3		
1				80 (>95:<5) ^b	
2				78 (1:2) ^c	
3				78 (<5:>95) ^b	
4				76 (<5:>95) ^b	
5				80 (1:1)	
6				79 (2:1) ^c	
7				81 (>95:<5) ^b	
8				82 (3:1) ^c	
9				82 (>95:<5) ^b	
10				80 (>95:<5) ^b	

^a All reactions were carried out on a 0.5 mmol scale using 1.2 equiv of NBS, 1.5 equiv of H₂O in toluene as solvent (0.2 M).

^b Crude product NMR did not reveal peaks for other isomers.

^c The isomers were inseparable by column chromatography.

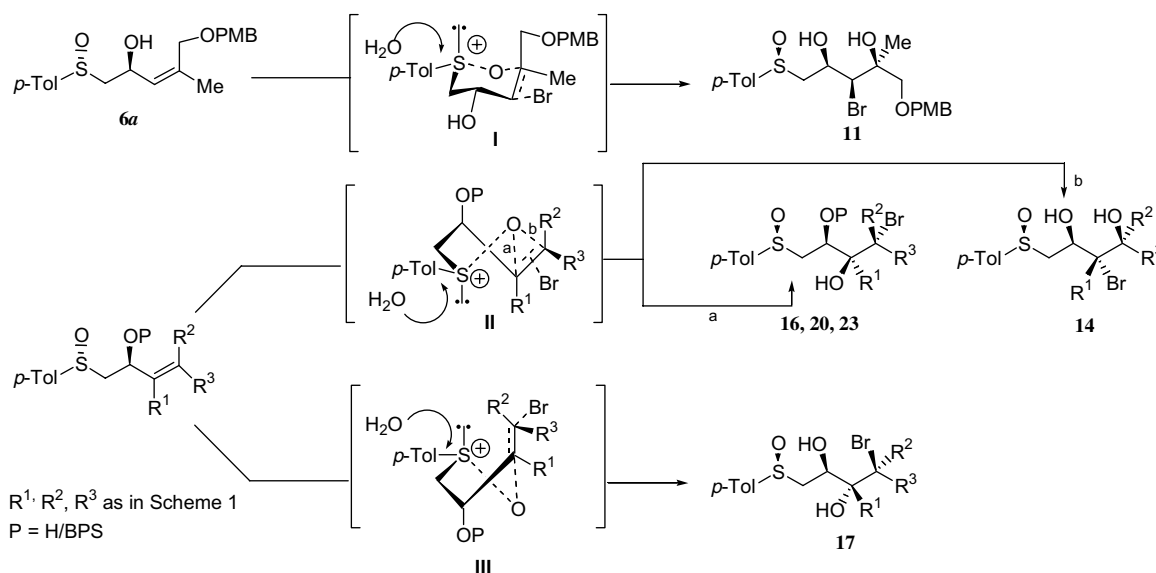
separated and subjected to a reaction with freshly recrystallized *N*-bromosuccinimide (NBS) in toluene in the presence of water to afford tertiary alcohols (Table 1) regio- and stereoselectively in good yield.

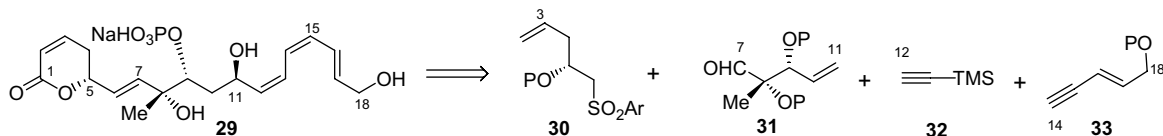
A perusal of the table reveals that the reaction is general, high yielding and obeys Markonikov's rule¹¹ to afford products regioselectively, wherein the nucleophile is linked to the more substituted carbon or the most electrophilic centre.¹² The structures were assigned by NMR (and NOE) study on the acetonides¹³ derived from diols **11**, **15**, **16** and **23**. Deprotection of the silyl groups in **20** followed by selective reprotection of the primary hydroxy group yielded **16**, thereby confirming the structure assigned to **20**. The enantiomeric relationship at C2–C4 between **14** and **15** was apparent after individually oxidizing them to the corresponding sulfones, whose ¹H NMR were identical. A similar oxidation proved the enantiomeric relationships at carbon atoms of **11/12**, **16/18** and **23/24**. The regio- and stereochemical outcome¹⁴ (illustrated only for *anti* series) of the reaction in entry 1 can be rationalized by invoking intermediate **I**, resulting from the rate-determining nucleophilic attack by the sulfinyl moiety onto a π -complexed bromonium ion, wherein A(1,3) interactions are minimized. Intermediate **II**, arising similarly would account for the only product in entries 3,7,9 and product **16** in entry 5. Product **17** in entry 5 could be arising via intermediate **III**. The strong destabilizing A(1,2) steric interactions that would be encountered between the siloxy and the methyl (*R*¹) substituent in **III** would account for the exclusive formation of **20** from **10a** via **II** (entry 7), Scheme 2.

The relative stereochemistry at sulfur and carbon atoms influences the stereoselectivity (compare entries 1/2, 5/6 and 7/8). By an appropriate choice of the *anti/syn*-isomers (compare entries 1/2 and 7/8) or by suitably protecting the allylic hydroxyl group (compare entries 5/7), the stereo-

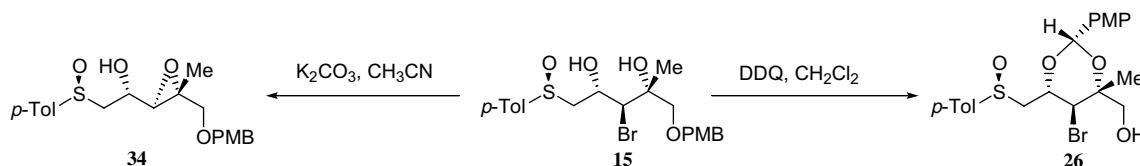
selectivity can be improved so as to get one isomer exclusively. *Syn* 1,2- and 1,3-diols with a quaternary stereogenic centre are thus readily obtained regio- and stereoselectively. It is worthwhile to note that by the reduction of the β -ketosulfoxide, obtained for instance, by reacting the lithium anion of methyl *p*-tolylsulfoxide with the esters corresponding to **2–5**, with DIBAL-H or DIBAL-H/ZnCl₂,¹⁵ the *anti*- or *syn*-isomers, respectively, can be obtained diastereoselectively. The presence of functional groups like bromine, sulfoxide and hydroxy groups provides a wide scope for manipulations, further carbon–carbon and carbon–heteroatom bond formations by intermolecular reactions. Also the bromohydrins can be readily transformed into chiral epoxides which are important intermediates,¹⁶ using either the secondary or tertiary hydroxy group. Also the hydroxymethyl and sulfinyl groups provide suitable handles for further elaboration at either ends of the molecule. As an illustration of the utility of the methodology we describe the synthesis of the C7–C11 fragment **31**, of fostriecin,¹⁷ Scheme 3. It was envisaged to construct (i) the C6–C7 double bond by a Julia olefination¹⁸ between subunits **30** and **31** and (ii) the lactone ring via a RCM reaction. In the resulting product, an aldehyde group at C11 was envisioned to be introduced by a selective hydroboration followed by oxidation. Further, the 1,3-asymmetric induction due to C9 substituent was to be capitalized on for introducing the C11 stereocentre selectively while forming the C11–C12 bond using **32**. Cadiotz–Chodkiewicz coupling¹⁹ using enyne **33** and further routine transformations were expected to furnish fostriecin. Compound **31** was envisioned to be derived from 1,3-diol **15**.

Epoxide formation by the selective participation of the secondary hydroxy group in **15** followed by the reductive elimination of sulfinyl moiety was expected to furnish the required allyl alcohol. In the event, the treatment of **15**²⁰

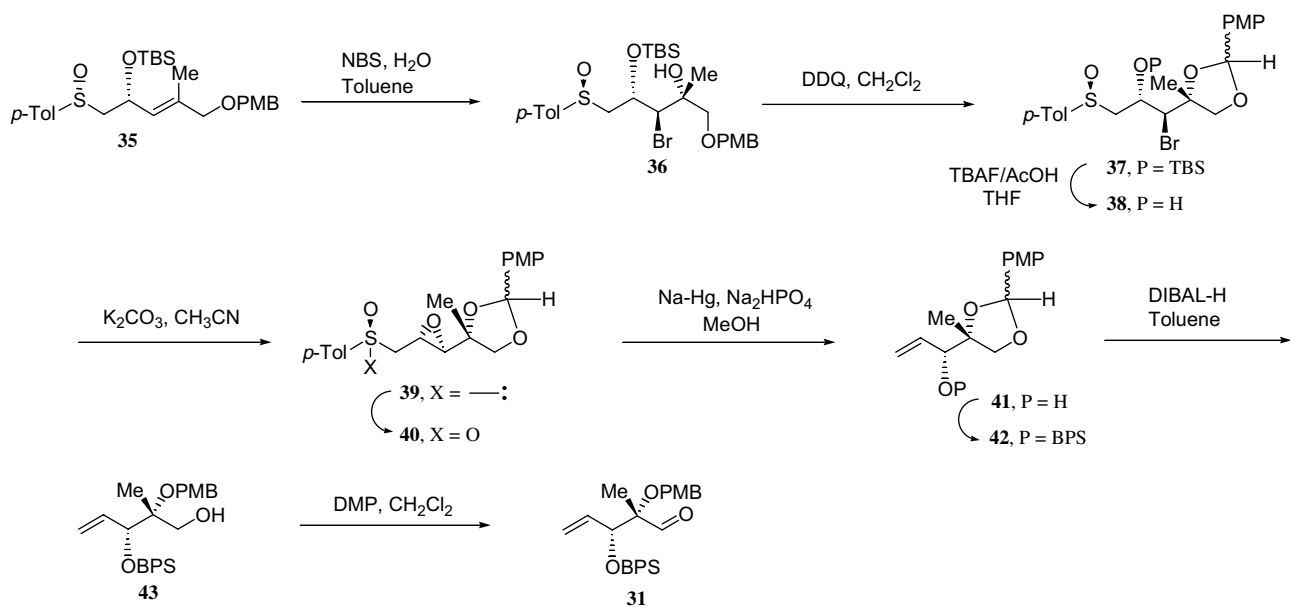




Scheme 3.



Scheme 4.



Scheme 5.

with anhydrous potassium carbonate in acetonitrile afforded epoxide **34** (89% yield) as the only product. The tertiary hydroxyl group therefore needed to be protected. Treatment of **15** with DDQ²¹ did not afford the 1,2-acetonide but the rearranged 1,3-acetonide **26** (76% yield) as the only product, **Scheme 4**.

Silyl ether **35** (90% yield) derived from **7s** was reacted with NBS to furnish bromohydrin **36** (76% yield, 9:1 dr, minor isomer not depicted). 1,2-Acetonide formation with DDQ proceeded without an incident to yield **37** as an epimeric mixture (75% yield). Deprotection of the silyl group using buffered TBAF and epoxide formation by base treatment yielded **39** (81% for two steps). Reductive elimination²² of epoxy sulfonamide proceeded less cleanly in comparison to the corresponding sulfone **40** to yield allyl alcohol **41** (64% overall yield), which was protected as its silyl ether **42** (92% yield). Regioselective reduction of the acetonide with DIBAL-H furnished primary alcohol **43**

as a 9:1 mixture of regioisomers (76% yield). Oxidation of **43** with DMP²³ proceeded cleanly to yield aldehyde **31** (95% yield), **Scheme 5**.

In conclusion, we have disclosed a highly regio- and stereoselective method for the preparation of 1,2- and 1,3-diols possessing a quaternary chiral centre. The methodology is general, high yielding and affords building blocks useful for the natural products synthesis. The synthesis of few natural products is presently underway and these would form subjects for later communications from our laboratory.

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Supplementary data

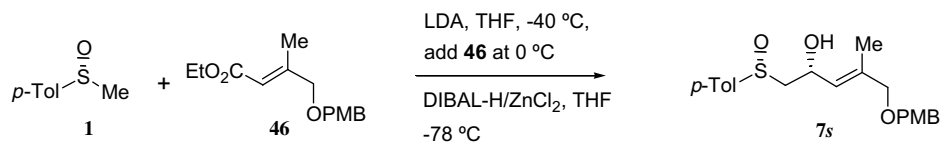
Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.tetlet.2007.12.046](https://doi.org/10.1016/j.tetlet.2007.12.046).

References and notes

- (a) Ramon, D. J.; Yus, M. Challenges and Solution for Organic Synthesis. In Christoffers, J., Baro, A., Eds.; Wiley-VCH, 2005. Chapter 8; (b) Christoffers, J.; Baro, A. *Angew. Chem., Int. Ed.* **2003**, *42*, 1688; (c) Denissova, I.; Barriault, L. *Tetrahedron* **2003**, *59*, 10105; (d) Corey, E. J.; Guzman-Perez, A. *Angew. Chem., Int. Ed.* **1998**, *37*, 388.
- Representative examples of tertiary carbinol containing bioactive compounds: Cinatrin C, Integerrimine, Harringtonine, K-252-a, Fostriecin, Viridofungin, Erythromycin and Zaragozic acid.
- Asymmetric epoxidation: (a) Katsuki, T. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Heidelberg, 1999; Vol. 2, Chapter 18.1 (b) Jacobsen, E. N.; Wu, M. F. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Heidelberg, 1999; Vol. 2, Chapter 18.2 (c) Murray, R. W. *Chem. Rev.* **1989**, *89*, 1187; Asymmetric hydroxylation: (d) Marko, I. E.; Svendsen, J. S. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Heidelberg, 1999; Vol. 2, Chapter 20 (e) Kolb, H. C.; Van Nieuwenhze, M. S.; Sharpless, K. B. *Chem. Rev.* **1994**, *94*, 2483; Biological methods: (f) Takagi, M.; Uemura, N.; Furuhashi, K. *Ann. N. Y. Acad. Sci.* **1990**, *613*, 697; (g) Subramanian, V. *J. Ind. Microbiol.* **1986**, *1*, 119; (h) Rossiter, J. T.; Williams, S. R.; Cass, A. E. G.; Ribbons, D. W. *Tetrahedron Lett.* **1987**, *28*, 5173.
- Jung, B.; Kang, S. H. *Proc. Natl. Acad. Sci. U.S.A.* **2007**, *104*, 1471.
- Cyanohydrin formation: (a) Fuerst, D. E.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2005**, *127*, 8964; (b) Riu, D. H.; Corey, E. J. *J. Am. Chem. Soc.* **2005**, *127*, 5384; (c) Tian, S.-K.; Hong, R.; Deng, L. *J. Am. Chem. Soc.* **2003**, *125*, 9990; (d) Chen, F.; Feng, X.; Qin, B.; Zhang, G.; Jiang, Y. *Org. Lett.* **2003**, *5*, 949, and references cited therein; Organometallic addition: (e) Forrat, V. J.; Prieto, O.; Ramon, D. J.; Yus, M. *Chem. Eur. J.* **2006**, *12*, 4431; (f) Chen, C.; Hong, L.; Xu, Z.-Q.; Zhao-Qing, V.; Liu, L.; Wang, R. *Org. Lett.* **2006**, *8*, 2277; (g) Canales, E.; Prasad, K. G.; Soderquist, J. A. *J. Am. Chem. Soc.* **2005**, *127*, 11572; (h) Li, H.; Walsh, P. J. *J. Am. Chem. Soc.* **2005**, *127*, 8355; Aldol reaction: (i) Moreau, X.; Bazan-Tezeda, B.; Campagne, J.-M. *J. Am. Chem. Soc.* **2005**, *127*, 7288; (j) Denmark, S. E.; Fan, Y. *J. Am. Chem. Soc.* **2002**, *124*, 4233; (k) Johnson, J. S.; Evans, D. A. *Acc. Chem. Res.* **2000**, *33*, 325.
- (a) Masaki, Y.; Arasaki, H.; Iwata, M. *Chem. Lett.* **2003**, *32*, 4; (b) Masaki, Y.; Arasaki, H.; Shiro, M. *Chem. Lett.* **2000**, *29*, 1180.
- For the oxidative functionalization of di-substituted alkenes refer: (a) Raghavan, S.; Rasheed, M. A.; Joseph, S. C.; Rajender, A. *Chem. Commun.* **1999**, 1845. For the application of this methodology in the synthesis of bioactive target molecules refer: (b) Raghavan, S.; Ramakrishna Reddy, S. *J. Org. Chem.* **2003**, *68*, 5754; (c) Raghavan, S.; Joseph, S. C. *Tetrahedron: Asymmetry* **2003**, *14*, 101; (d) Raghavan, S.; Rasheed, M. A. *Tetrahedron: Asymmetry* **2003**, *14*, 1371; (e) Raghavan, S.; Rajender, A.; Yadav, J. S. *Tetrahedron: Asymmetry* **2003**, *14*, 2093; (f) Raghavan, S.; Joseph, S. C. *Tetrahedron Lett.* **2003**, *44*, 6713; (g) Raghavan, S.; Rajender, A. *J. Org. Chem.* **2003**, *68*, 7094; (h) Raghavan, S.; Rasheed, M. A. *Tetrahedron* **2003**, *59*, 10307; (i) Raghavan, S.; Joseph, S. C. *Tetrahedron Lett.* **2003**, *44*, 8237; (j) Raghavan, S.; Sreekanth, T. *Tetrahedron: Asymmetry* **2004**, *15*, 565; (k) Raghavan, S.; Tony, K. A. *Tetrahedron Lett.* **2004**, *45*, 2639; (l) Raghavan, S.; Ramakrishna Reddy, S. *Tetrahedron Lett.* **2004**, *45*, 5593.
- Drago, C.; Caggiano, L.; Jackson, R. F. W. *Angew. Chem., Int. Ed.* **2005**, *44*, 7221.
- The unsaturated aldehydes were obtained from the corresponding esters in two steps, (a) reduction using DIBAL-H to furnish the allyl alcohols followed by (b) oxidation using Swern protocol. The esters themselves were prepared using procedures reported in the literature, ester corresponding to **2**, refer: Koppisch, A. T.; Blagg, B. S. J.; Poulter, C. D. *Org. Lett.* **2000**, *2*, 215. For ester corresponding to **3** refer: Baldwin, I. R.; Whitby, R. J. *Chem. Commun.* **2003**, 2786; For ester corresponding to **4** refer: Nicolaou, K. C.; Liu, J.-J.; Yang, Z.; Ueno, H.; Sorensen, E. J.; Claiborne, C. F.; Guy, R. K.; Hwang, C.-K.; Nakada, M.; Nantermet, P. G. *J. Am. Chem. Soc.* **1995**, *117*, 634; For ester corresponding to **5** refer: Marshall, J. A.; Trometer, J. D.; Blough, B. E.; Crute, T. D. *J. Org. Chem.* **1988**, *53*, 4274.
- The configuration of *anti*- and *syn*-isomers were assigned by comparison of coupling constants for the methylene protons directly bonded to the carbon of the β -hydroxy sulfoxide moiety and the chemical shift of the β -methine proton. Refer: Carreno, M. C.; Garcia Ruano, T. L.; Martin, A. M.; Pedregal, C.; Rodrigues, J. H.; Rubio, A.; Sanchez, J.; Solladie, G. *J. Org. Chem.* **1990**, *55*, 2120; and references cited therein.
- (a) Markovnikov, V. *Liebigs Ann. Chem.* **1870**, *153*, 256. For a modern interpretation, see: (b) Isenberg, N.; Grdinic, M. *J. Chem. Educ.* **1969**, *46*, 601.
- De la Mare, P. B. D.; Bolton, O. *Electrophilic Addition to Unsaturated Systems*; Elsevier: Amsterdam, 1982.
- See Electronic Supplementary data.
- The inversion of sulfoxide configuration expected from neighbouring group participation has been conclusively proven in earlier work by labelling studies using $H_2^{18}O$ and X-ray crystallography, refer: (a) Raghavan, S.; Ramakrishna Reddy, S.; Tony, K. A.; Naveen Kumar, Ch.; Varma, A. K.; Nangia, A. *J. Org. Chem.* **2002**, *67*, 5838; (b) Raghavan, S.; Rajender, A.; Joseph, S. C.; Rasheed, M. A.; Ravikumar, K. *Tetrahedron: Asymmetry* **2004**, *15*, 365; (c) Raghavan, S.; Naveen Kumar, Ch.; Tony, K. A.; Ramakrishna Reddy, S.; Ravikumar, K. *Tetrahedron Lett.* **2004**, *45*, 7231; Ref. 7l. Also refer: (d) Dalton, D. R.; Dutta, V. P.; Jones, D. C. *J. Am. Chem. Soc.* **1968**, *90*, 5498. The positive optical rotation for the substrates and the negative rotation for the products also support inversion of sulfoxide configuration, refer: Electronic Supplementary data.
- (a) Solladie, G.; Demailly, G.; Greck, C. *Tetrahedron Lett.* **1985**, *26*, 435; (b) Solladie, G.; Frechou, C.; Demailly, G.; Greck, C. *J. Org. Chem.* **1986**, *51*, 1912; (c) Carreno, M. C.; Garcia Ruano, J. L.; Martin, A.; Pedregal, C.; Rodrigues, J. H.; Rubio, A.; Sanchez, J.; Solladie, G. *J. Org. Chem.* **1990**, *55*, 2120.
- (a) Zhao, H.; Engers, D. W.; Morales, C. L.; Pagenkopf, B. L. *Tetrahedron* **2007**, *63*, 8774; (b) Lautens, M.; Maddess, M. L.; Sauer, E. L. O.; Ouellet, S. G. *Org. Lett.* **2002**, *4*, 83; (c) Jung, M. E.; Lee, C. P. *Org. Lett.* **2001**, *3*, 333; (d) Jung, M. E.; Lee, W. S.; Sun, D. *Org. Lett.* **1999**, *2*, 307; (e) Jung, M. E.; D'Amico, D. C. *J. Am. Chem. Soc.* **1997**, *119*, 12150; (f) Jung, M. E.; Marquez, R. *Tetrahedron Lett.* **1999**, *40*, 3129; (g) Shimazaki, M.; Hara, H.; Suzuki, K.; Tsuchihashi, G. *Tetrahedron Lett.* **1987**, *28*, 5891; (h) Maruoka, K.; Ooi, T.; Nagahara, S.; Yamamoto, H. *Tetrahedron* **1991**, *47*, 6983.
- (a) Boger, D. L.; Ichikawa, S.; Zhong, W. *J. Am. Chem. Soc.* **2001**, *123*, 4161; (b) Chavez, D. E.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **2001**, *40*, 3667; (c) Cossy, J.; Pradaux, F.; BouzBouz, S. *Org. Lett.* **2001**, *3*, 2233; (d) Reddy, Y. K.; Falck, J. R. *Org. Lett.* **2002**, *4*, 969; (e) Wang, Y. G.; Kobayashi, Y. *Org. Lett.* **2002**, *4*, 4615; (f) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. *Chem. Commun.* **2002**, 742; (g) Esumi, T.; Okamoto, N.; Hatakeyama, S. *Chem. Commun.* **2002**, 3042; (h) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. *J. Am. Chem. Soc.* **2003**, *125*, 8238; (i) Fujii, K.; Maki, K.; Kanai, M.; Shibasaki, M. *Org. Lett.*

2003, 5, 733; (j) Trost, B. M.; Frederiksen, M. U.; Papillon, J. P. N.; Harrington, P. E.; Shin, S.; Shireman, B. T. *J. Am. Chem. Soc.* **2005**, *127*, 3666.

20. Compound **15** was obtained from **7s** prepared in two steps (i. formation of β -keto sulfoxide and ii. stereoselective reduction of keto sulfoxide, 53% overall yield) as depicted.



18. Blakemore, P. R.; Cole, W. J.; Kocienski, P. J.; Morley, A. *Synlett* **1998**, 26–28.

19. (a) Cadiotz, P.; Chodkiewicz, W. In *Chemistry of Acetylenes*; Viehe, H. G., Ed.; Marcel Dekker: New York, 1969; pp 597–647; (b) Chodkiewicz, W. *Ann. Chim. (Paris)* **1957**, *2*, 819; (c) Siemsen, P.; Livingstone, R. C.; Diedrich, F. *Angew. Chem., Int. Ed.* **2000**, *39*, 2632.

21. Hanessian, S.; Cooke, N. G.; DeHoff, B.; Sakito, Y. *J. Am. Chem. Soc.* **1990**, *112*, 5276.

22. Trost, B. M.; Arndt, H. C.; Strege, P. E.; Verhoeven, T. R. *Tetrahedron Lett.* **1976**, 3477.

23. Dess, D. B.; Martin, J. C. *J. Org. Chem.* **1983**, *48*, 4155.